

College Chemistry II
CH 122 – Spring 2021**Lecture CRN: 43436** Mondays: 10:30 – 11:50 AM (via Zoom)**Lab CRN: 43648** Remote (Lab materials will be uploaded to Moodle every week)**Lab CRN: 43649** Remote (Lab materials will be uploaded to Moodle every week)**Instructor:** Dr. David Rogow**Contact:** rogowd@linnbenton.edu**Office Hours:** Wednesdays 1 – 3 pm (via Zoom)**Course Information:**

The second of a three term college chemistry sequence for students in human performance, certain health occupations programs, agriculture, animal science, and fisheries and wildlife. This sequence is for students who have had no previous training in chemistry and whose program of study requires only a one-year sequence of college chemistry. Topics include thermochemistry, gases, liquids, solids, intermolecular forces, solutions, chemical kinetics, and chemical equilibrium.

Student Learning Outcomes:

1. Solve scientific problems with quantitative methods regarding thermochemistry, gases, molarity, bonding, phase changes, colligative properties, chemical kinetics, and chemical equilibrium.
2. Apply chemical principles related to intermolecular attractions of pure substances and solutions, properties of gas, liquid, and solid, phase transitions, colligative properties, chemical kinetics, and chemical equilibrium.
3. Work safely in a laboratory environment while observing and accurately recording measurements related to chemical phenomena.

Minimum Requirements:

Prerequisites: MTH 111 College Algebra and CH 121 College Chemistry with a grade of "C" or better. Corequisite: CH 122L College Chemistry II Lab.

Workload Expectation:

The college has determined that a student taking a chemistry course spends a minimum of 3 – 4 hours of work per week outside of class for every credit hour for self-study. Examples of outside work include writing lab reports, reading, reviewing lecture materials, study time, working practice problems, and doing homework assignments.

Required Course Materials (Available for you to purchase at the bookstore):

1. Access Code for Knewton Alta Online Homework (The access code from last term works.)
2. Chemistry 122 Lecture Manual (Students order a hard-copy from the LBCC book store)
3. Non-graphing/non-programmable Scientific Calculator (TI 30xa). Students will be required to use a non-graphing/non-programmable scientific calculator for quizzes and/or exams.

Course Format: Lecture will be synchronous and the lab will be asynchronous. Synchronous means you are required to attend the virtual lecture (Mondays 10:30 – 11:50 am) via Zoom. However, the lab will be asynchronous, meaning you will work on each lab during the week on your own time.

Attendance and Classroom Decorum:

Class attendance and participation are very important to be successful in the learning of chemistry. Students are encouraged to attend class regularly, on time, and engage in activities and/or discussions. This includes online Zoom meetings as a replacement for lecture.

Grade Assessments:

Your grade will be assigned based on your performance in the following areas:

Laboratory Exercises	8 × 10 pts.	= 80 pts.	(20 %)
Knewton online Homework	6 × 10 pts.	= 60 pts.	(15 %)
Learning Assessments	5 × 50 pts.	= 250 pts.	(65 %)

Total	390 pts.
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Course Grade:

Assignment of course grades will follow an approximate breakdown of

- A = 90-100% Excellent Work
- B = 80-89% Good Work
- C = 70-79% Average Work
- D = 60-69% Poor Work
- F = 0-59% Failing Work

An incomplete grade (I) may be given at the discretion of the instructor. However, a student must have a passing grade at the time an incomplete is assigned.

Learning Assessments (LAs):

The LAs are designed to evaluate student's understanding of the materials that they have learned from the previous weeks. Each LA is worth 50 points. A total of 250 points can be earned from LAs.

Exam Policies:

All exams/assessments are taken on Moodle. Exams must be taken within the scheduled time unless prior arrangement is made. Students who have conflicts with exam days due to other College functions, illness, or family emergencies must contact the instructor prior to the exam. Documentation of the College function, illness and/or family emergency must be provided to schedule a make-up exam. Any academic dishonesty during any exams including cheating, posting questions on answer websites (Chegg, CourseHero, etc...), or using materials that haven't been approved, will result in a score of ZERO for the exam, and possibly further disciplinary action.

The final exam is not comprehensive (only covers chapters 12 & 13). A missed final exam will receive a score of zero.

Laboratory Exercise

The laboratory experience is a vital part of this course. The labs are remote and the videos/data for each lab will be posted every week to Moodle. Students are expected to complete each laboratory at the scheduled time. **You must receive at least 70% of the total lab points in order to pass the course regardless of passing the lecture. Also, if you miss more than three labs or turn in fewer than five reports you will not receive a passing grade for the course.** The deadlines will not be extended so please plan accordingly! The report should be submitted to the lab instructor through Moodle. The report must be in PDF format, which can be accomplished by converting a Microsoft Word document into a .pdf, or using 'Adobe Scan', 'CamScanner', 'Google Drive' or any other free app for making PDFs. The text of the report can be typed if you like. You can either hand-write your calculations or use an equation editor. Please do not type equations using normal text.

Online Homework:

To succeed in chemistry, like learning a foreign language, you should study and practice every day. As material is covered you will find the problems are easier to work and not as time consuming as if they are attempted just before the due date. Refer to the schedule for homework due dates. You can access **Knewton Alta Online Homework** via Moodle. Each homework assignment is worth 10 points. Homework is due by 11:59 pm on the dates listed in the lecture schedule.

NOTE: This homework is adaptive to each learner. If you don't get consecutive correct answers, the system will think that you have not mastered a particular topic; therefore, it will throw more problems at you. Be sure to start early and not try to do the online homework right before it is due to avoid frustration; work on it early and often.

For late homework, students can turn in completed assignments after the due date up to 2 days late. However, there will be a 5% penalty per day late.

For your first time doing homework, the Knewton Alta Online Homework will prompt you to enter an access code. You can purchase this access code online or at the LBCC bookstore. Knewton Alta offers a grace period on payment; for most courses, this is 14 days from the first day of the term. You can also pay when you click on the assignment.

Resources:

Your success is very important to me! I encourage you to seek help from one or more of the following resources:

1. Instructor office hours (see the front page for days, times and locations)
2. Ask questions during lecture
3. Science Help Desk
4. Academic Support (<http://linnbenton.edu/future-students/academic-support/>)

Science Help Desk:

If you need help in any physics or chemistry course, you should “drop by” the Physical Science Desk, now offered online! The Help Desk is staffed approximately 20 hours per week. Please visit the Science Help Desk web page ([click here](#)) for details and hours.

Roadrunner Resource Center for Basic Needs:

Any student who has difficulty affording tuition, course materials, hygiene materials, food, who lacks a safe and stable place to live, who needs transportation, and believes this may affect their performance in the course, is urged to contact the [Roadrunner Resource Center](#) for support (Resources@linnbenton.edu).

Center for Accessibility Resources:

LBCC is committed to inclusiveness and equal access to higher education. If you have approved accommodations through the Center for Accessibility Resources (CFAR) and would like to use your accommodations in the class, please talk to your instructor as soon as possible to discuss your needs. If you believe you may need accommodations but are not yet registered with CFAR, please visit the [CfAR Website](#) for steps on how to apply for services or call [\(541\) 917-4789](tel:5419174789).

Tips for Success:

- Attend every lecture, and come prepared!
- Review lecture notes after lecture; clarify confusing points immediately; use your notes to guide your studying
- Doing homework problems regularly
- Form a study group; take turns “teaching” each other concepts/problems
- Repeat homework and/or worksheet problems until you can do them quickly, without looking at any notes or answer keys

Drop/Withdraw Policy:

If you are withdrawing from the class you must file a Schedule Change Form with Registration or use WebRunner. If you formally drop the class by Monday of the second week of the term, you will receive a tuition refund. If you withdraw after the Monday of the second week of instruction through the seventh week a 'W' will show up on your transcript. No withdrawals are allowed after the end of the seventh week. An instructor may not assign a "W" grade.

If you received financial aid or veteran's benefits PLEASE talk with associates at the appropriate office to determine what effects on eligibility dropping a course will have. Don't jeopardize your eligibility!! You can contact the Financial Aid Office by calling (541) 917-4850 or by visiting the Financial Aid Office in Takena Hall.

If you stop attending the course without formally withdrawing you will continue to accumulate grades (zeroes for all assignments not turned in) and will receive the grade earned. You will also be held accountable for all charges on your account.

Academic Integrity:

"An instructor has the right to issue a grade of F for the course in which the instructor has reason to believe the student has cheated. A student has the right to appeal such action in accordance with the [Students' Rights, Responsibilities and Conduct Policy](#)." The preceding statement is Administrative Rule No. 7030-02.

LBCC Comprehensive Statement of Nondiscrimination:

LBCC prohibits unlawful discrimination based on race, color, religion, ethnicity, use of native language, national origin, sex, sexual orientation, gender, gender identity, marital status, disability, veteran status, age, or any other status protected under applicable federal, state, or local laws. For further information see Board Policy P1015 in our Board Policies and Administrative Rules.

Changes to the Syllabus:

The instructor reserves the right to change the contents of this syllabus due to unforeseen circumstances. You will be given notice of relevant changes in class, through a Moodle Announcement, or through LBCC e-mail.

Course Content**Chapter 8 – Thermochemistry**

- 8.1 Nature of Energy and the First Law of Thermodynamics
- 8.2 Thermal Energy, Temperature, and Heat
- 8.3 Measuring Heat
- 8.4 Enthalpy - Heat of Reaction (ΔH_{rxn})
- 8.5 Measuring Enthalpy from Calorimetry
- 8.6 Measuring Enthalpy from Hess's Law
- 8.7 Measuring Enthalpy from Standard Enthalpies of Formation

Chapter 9 – Gases

- 9.1 Pressure
- 9.2 Simple Gas Laws
- 9.3 The Kinetic Molecular Theory
- 9.4 The Ideal Gas Law and Its Applications
- 9.5 Mixtures of Gases and Partial Pressure
- 9.6 Gas Stoichiometry

Chapter 10 – Solids, Liquids & Their Intermolecular Forces

- 10.1 Intermolecular Forces
- 10.2 Properties of Liquids
- 10.3 Phase Transitions
- 10.4 Phase Diagrams

Chapter 11 – Solutions

- 11.1 Types of Solutions
- 11.2 Formation of Solutions and Dissolution Process
- 11.3 Electrolytes
- 11.4 Factors that Affect Solubility
- 11.5 Concentration Units
- 11.6 Dilution of Solution
- 11.7 Colligative Properties

Chapter 12 – Chemical Kinetics

- 12.1 Chemical Reaction Rate
- 12.2 Factors Affecting Reaction Rate
- 12.3 Collision Theory
- 12.4 Differential Rate Laws
- 12.5 Integrated Rate Laws
- 12.6 Reaction Mechanisms

Chapter 13 – Chemical Equilibrium

- 13.1 Concept of Dynamic Equilibrium
- 13.2 Reaction Quotient (Q) and Equilibrium Constant (K)
- 13.3 Chemical Equations and Equilibrium Constant Relation
- 13.4 Equilibrium Calculations
- 13.5 Le Chatelier's Principle

Course Schedule

**Note: This schedule of topics, homework due dates, and exam dates is subject to change. All homework assignments are due by 11:59 pm on the date indicated in the lecture schedule.

Week No./Dates	Textbook Sections/Learning Assessments (LAs)		Assignments/Due Dates
1 (3/29-4/2)	Introduction, Syllabus, 8.1 – 8.5		<i>All assignments are due by 11:59 pm on the due date</i>
2 (4/5-4/9)	8.6 – 8.7, 9.1 – 9.2	LA1 (CH-8) 4/8-4/9	Lab 1: Safety Activities (Due 4/14) CH-8 HW due (Due 4/7)
3 (4/12-4/16)	9.3 – 9.5		Lab 2: Energy Content of Fuel (Due 4/21)
4 (4/19-4/23)	9.5 – 9.6, 10.1	LA2 (CH-9) 4/22-4/23	Lab 3: Universal Gas Constant (Due 4/28) CH-9 HW due (Due 4/21)
5 (4/26-4/30)	10.2 – 10.4, 11.1 – 11.2		Lab 4: Chromatography of Juice (Due 5/5)
6 (5/3-5/7)	11.3 – 11.6	LA3 (CH-10) 5/6-5/7	Lab 5: Instant C&H Packs (Due 5/12) CH-10 HW due (Due 5/5)
7 (5/10-5/14)	11.7		Lab 6: Colligative Properties (Due 5/19)
8 (5/17-5/21)	12.1 – 12.3	LA4 (CH-11) 5/20-5/21	Lab 7: Iodine Clock Reaction (Due 5/26) CH-11 HW due (Due 5/19)
9 (5/24-5/28)	12.4 – 12.6		Lab 8: Review Worksheet (Due 6/2)
10 (5/31-6/4)	Memorial Day Holiday Monday, no class this week 13.1 – 13.5		<i>No Lab this week</i> CH-12 HW due (Due 6/2) CH-13 HW due (Due 6/7)
11 (6/7-6/11)	LA5: Final Exam: Chapters 12 & 13 Available 6/8-6/9		